

## Prostanoids. LXXVIII.\* New $sp^2$ -Functionalized 4-Hydroxy-2-cyclopentenones from 6,7-Dichloro-1,4-dioxaspiro[4.4]non-6-en-8-one

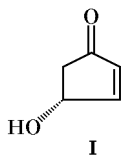
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**Abstract**—Readily accessible 6,7-dichloro-1,4-dioxaspiro[4.4]non-6-en-8-one was converted into 2,3-dichloro-, 2-ethylthio-, and 3-chloro-2-ethylthio-4-hydroxy-2-cyclopentenones.

Protected derivatives of (4R)-4-hydroxy-2-cyclopentenones (**I**) are widely used in the synthesis of prostanoids, following the triply convergent scheme [2–4]. In the present communication we report on the synthesis of new 4-hydroxycyclopentenones on the basis of readily accessible 6,7-dichloro-1,4-dioxaspiro[4.4]non-6-en-8-one (**II**) [5]. Products **III–V** attract interest as potential starting compounds for alternative approaches (other than triply convergent synthesis) to vicinal 2,3-dialkyl derivatives of 4-hydroxy-2-cyclopentenones.



The transformations of compound **II** into products **III–V** are shown in Scheme 1. The reduction of **II** with  $\text{LiAlH}_4$  gave alcohol **VI** which was treated with 15% hydrochloric acid in  $\text{Me}_2\text{CO}$  to obtain the desired 2,3-dichloro-4-hydroxy-2-cyclopentenone (**III**). The proposed procedure for preparation of 2-ethylthio derivative **V** is also simple. It includes nucleophilic substitution of the chlorine atom in **II**, exhaustive reduction of ketone **VII** to alcohol **VIII** with  $\text{LiAlH}_4$ , and acid hydrolysis to remove the acetal protection.

As a result, we isolated compound **IV** and ether **IX** as by-product in 52 and 26% yield, respectively

(calculated on ethylthio derivative **VII**. Unlike lithium aluminum hydride, the reduction of **VII** with sodium tetrahydridoborate gave alcohol **X** in which the vinyl chlorine atom remained intact. Mild acid hydrolysis of **X** smoothly afforded 2,3-disubstituted hydroxycyclopentenone **V**. Presumably, extremely facile hydrolysis of the dioxolane moiety in compounds **IV**, **VIII**, and **X** is favored by intramolecular assistance by the hydroxy group (cf. [6, 7]).

### EXPERIMENTAL

The IR spectra were recorded on a UR-20 instrument from samples prepared as thin films. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were obtained on a Bruker AM-300 spectrometer at 300 and 75.47 MHz, respectively;  $\text{CDCl}_3$  was used as solvent, and TMS, as internal reference.

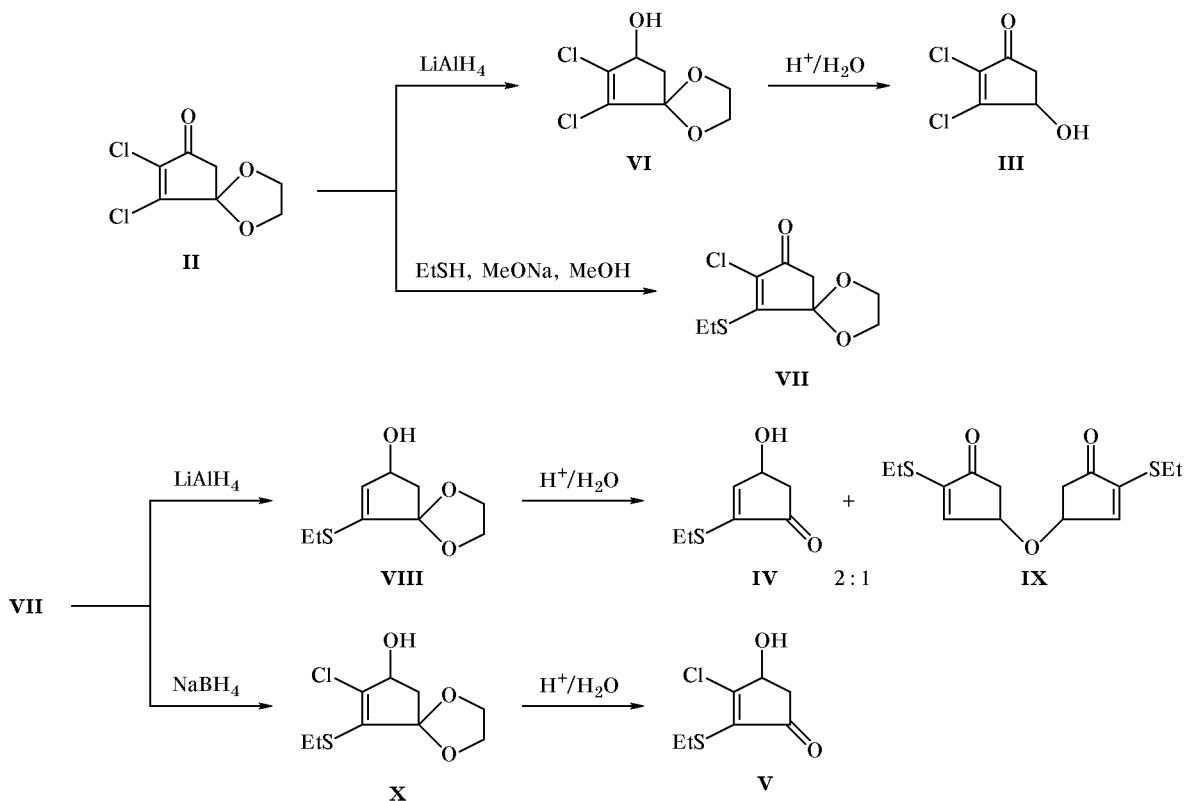
**2,3-Dichloro-4-hydroxy-2-cyclopentenone (III).** To a solution of 0.11 g (0.5 mmol) of compound **VI** in 10 ml of acetone we added 0.1 ml of 15% hydrochloric acid. The mixture was kept for 30 min at  $20^\circ\text{C}$  and evaporated, and the residue was purified by chromatography on silica gel to obtain 60 mg (82%) of ketone **III**. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1620, 1745, 3400.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 3.13 d.d (1H, 5-H,  $J = 18.6, 6.4$  Hz). 2.68 d.d (1H, 5-H,  $J = 18.6, 1.7$  Hz), 4.0–4.25 m (1H, OH), 4.91 d.d (1H, 4-H,  $J = 1.7, 6.4$  Hz).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 43.54 ( $\text{C}^5$ ), 69.76 ( $\text{C}^4$ ), 134.03 ( $\text{C}^2$ ), 163.81 ( $\text{C}^3$ ), 194.66 ( $\text{C}=\text{O}$ ).

**2-Ethylthio-4-hydroxy-2-cyclopentenone (IV) and bis(4-oxo-3-ethylthio-2-cyclopentenyl) ether (IX).** To a solution of 120 mg of compound **VIII**

\* For communication LXXVII, see [1].

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Scheme 1.



in 10 ml of acetone we added 0.1 ml of 15% hydrochloric acid. The mixture was stirred for 30 min at 20°C and evaporated, and the residue was subjected to chromatography on silica gel to isolate 60 mg (52%) of ketone **IV** and 30 mg (26%) of ether **IX**.

**2-Ethylthio-4-hydroxy-2-cyclopentenone (IV)**. Oily liquid. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1604, 1712, 3424.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.35 t (3H,  $\text{CH}_3$ ,  $J = 7.4$  Hz), 2.00–2.30 m (1H, OH), 2.41 d.d (1H,  $\text{CH}_2$ ,  $J = 18.7, 2.0$  Hz), 2.86 q (2H,  $\text{SCH}_2$ ,  $J = 7.4$  Hz), 2.92 d.d (1H,  $\text{CH}_2$ ,  $J = 18.7, 6.1$  Hz), 5.00–5.04 m (1H, 4-H), 6.93 d (1H, 3-H,  $J = 2.6$  Hz).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 13.56 ( $\text{CH}_3$ ), 24.99 ( $\text{SCH}_2$ ), 44.96 ( $\text{C}^5$ ), 69.16 ( $\text{C}^4$ ), 145.03 ( $\text{C}^2$ ), 148.65 ( $\text{C}^3$ ), 202.39 ( $\text{C}=\text{O}$ ).

**3-Chloro-2-ethylthio-4-hydroxy-2-cyclopentenone (V)** was synthesized by hydrolysis of compound **X**, following the procedure described above for ketone **IV**. Oily substance. Yield 84%. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1570, 1712, 3416.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.25 t (3H,  $\text{CH}_3$ ,  $J = 7.4$  Hz), 2.57 d.d (1H,  $\text{CH}_2$ ,  $J = 18.5, 2.1$  Hz), 2.85–3.15 m (1H, OH), 2.98 d.d (1H,  $\text{CH}_2$ ,  $J = 18.5, 6.5$  Hz), 3.15 q (2H,  $\text{SCH}_2$ ,  $J = 7.4$  Hz), 4.73 d.d (1H, 4-H,  $J = 6.5, 2.1$  Hz).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 15.60 ( $\text{CH}_3$ ), 24.37 ( $\text{SCH}_2$ ),

44.71 ( $\text{C}^5$ ), 70.62 ( $\text{C}^4$ ), 137.63 ( $\text{C}^2$ ), 162.70 ( $\text{C}^3$ ), 198.02 ( $\text{C}=\text{O}$ ).

**6,7-Dichloro-1,4-dioxaspiro[4.4]non-6-en-8-ol (VI)**. A solution of 0.42 g (2 mmol) of ketone **II** in 10 ml of THF was added dropwise with vigorous stirring at room temperature to a suspension of 0.08 g of  $\text{LiAlH}_4$  in 10 ml of THF. The mixture was stirred for 15 min and diluted with 30 ml of diethyl ether, and 1 ml of a 10% aqueous solution of KOH was added. The organic phase was dried over  $\text{MgSO}_4$  and evaporated, and the residue was subjected to chromatography on silica gel using 4:1 pentane–ethyl acetate as eluent. Product **VI** was isolated as an oily substance. Yield 0.35 g (83%). IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 1636, 3440.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 2.09 d.d (1H, 9-H,  $J = 14.2, 3.5$  Hz), 2.60 d.d (1H, 9-H,  $J = 14.2, 7.3$  Hz), 3.70–3.90 m (1H, OH), 3.90–4.25 m (4H,  $\text{CH}_2\text{O}$ ), 4.50–4.60 m (1H, 8-H).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 44.45 ( $\text{C}^9$ ), 65.70 ( $\text{CH}_2\text{O}$ ), 66.03 ( $\text{CH}_2\text{O}$ ), 71.54 ( $\text{C}^8$ ), 112.50 ( $\text{C}^5$ ), 132.57 ( $\text{C}^6$ ), 134.13 ( $\text{C}^7$ ).

**7-Chloro-6-ethylthio-1,4-dioxaspiro[4.4]non-6-en-8-one (VII)**. To a solution of sodium methoxide, prepared from 0.03 g (1.3 mmol) of metallic sodium and 5 ml of anhydrous methanol, we added with stirring 0.81 g (1.3 mmol) of ethanethiol and 0.3 g

(1.2 mmol) of ketone **II**. The mixture was stirred for 0.5 h at 20°C, 10 ml of water was added, and the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 20 ml). The combined extracts were washed with a saturated aqueous solution of sodium chloride until neutral reaction, dried over MgSO<sub>4</sub>, and evaporated to obtain 0.37 g of a crude product which was purified by column chromatography on silica gel using 2:1 pentane–ethyl acetate as eluent. Yield 0.27 g (81%), oily substance. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1570, 1720. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.33 t (3H, CH<sub>3</sub>,  $J$  = 7.1 Hz), 2.70 s (2H, 9-H<sub>2</sub>), 3.40 m (2H, SCH<sub>2</sub>), 4.00–4.20 m (4H, CH<sub>2</sub>O). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 14.97 (CH<sub>3</sub>), 25.16 (SCH<sub>2</sub>), 65.77 (CH<sub>2</sub>O), 110.51 (C<sup>5</sup>), 130.99 (C<sup>7</sup>), 164.73 (C<sup>6</sup>), 191.94 (C=O). Found, %: C 45.80; H 4.58; Cl 15.40; S 13.40. C<sub>9</sub>H<sub>11</sub>ClO<sub>3</sub>S. Calculated, %: C 46.06; H 4.72; Cl 15.11; S 13.66.

**6-Ethylthio-8-hydroxy-1,4-dioxaspiro[4.4]non-6-ene (VIII)** was synthesized by reduction of ketone **VII**, following the procedure described above for compound **VI**. Oily substance, yield 83%. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1610, 3450. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.33 t (3H, CH<sub>3</sub>,  $J$  = 7.4 Hz), 1.97 d.d (1H, CH<sub>2</sub>,  $J$  = 14.0, 3.2 Hz), 2.53 d.d (1H, CH<sub>2</sub>,  $J$  = 14.0, 4.8 Hz), 2.30–2.50 m (1H, OH), 2.79 q (2H, SCH<sub>2</sub>,  $J$  = 7.4 Hz), 3.90–4.13 m (4H, CH<sub>2</sub>O), 4.65–4.78 m (1H, 8-H), 5.62 d (1H, 7-H,  $J$  = 2.5 Hz). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 13.47 (CH<sub>3</sub>), 25.12 (SCH<sub>2</sub>), 46.74 (C<sup>9</sup>), 65.71 (CH<sub>2</sub>O), 65.92 (CH<sub>2</sub>O), 72.21 (C<sup>8</sup>), 117.02 (C<sup>5</sup>), 128.06 (C<sup>7</sup>), 145.37 (C<sup>6</sup>).

**Bis(3-ethylthio-4-oxo-2-cyclopentenyl) ether (IX)**. Oily liquid. IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1572, 1712. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.35 t and 1.37 t (6H, CH<sub>3</sub>,  $J$  = 7.4 Hz), 2.47 d.d and 2.50 d.d (2H, 5-H, 5'-H,  $J$  = 18.4, 1.8 Hz), 2.70–3.00 m (6H, CH<sub>2</sub>S, 5-H, 5'-H), 4.75–4.90 m (2H, 4-H, 4'-H), 6.89 d and 6.95 d (2H, 3-H, 3'-H,  $J$  = 2.6, 2.8 Hz). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 13.60 (CH<sub>3</sub>), 25.05 (SCH<sub>2</sub>), 42.76 (C<sup>5</sup>), 74.90 and 75.25 (C<sup>4</sup>, C<sup>4'</sup>), 145.52 and 145.64 (C<sup>2</sup>, C<sup>2'</sup>), 146.59 and 146.63 (C<sup>3</sup>, C<sup>3'</sup>), 201.09 (C=O).

**7-Chloro-6-ethylthio-8-hydroxy-1,4-dioxaspiro[4.4]non-6-ene (X)**. To a solution of 0.46 g (2 mmol)

of ketone **VII** in 10 ml of ethanol we added at 0°C 0.03 g (0.8 mmol) of sodium tetrahydridoborate. The mixture was stirred for 1 h, 3 ml of acetone was added to decompose excess reducing agent, and the mixture was evaporated. The residue was dissolved in 10 ml of water, the solution was extracted with methylene chloride (3 × 20 ml), the extract was dried over MgSO<sub>4</sub> and evaporated, and the residue was subjected to chromatography on silica gel using 4:1 pentane–ethyl acetate as eluent. Yield 0.45 g (96%). IR spectrum,  $\nu$ , cm<sup>-1</sup>: 1570, 1712, 3430. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 1.19 t (3H, CH<sub>3</sub>,  $J$  = 7.4 Hz), 2.01 d.d (1H, CH<sub>2</sub>,  $J$  = 14.0, 3.8 Hz), 2.01 q (2H, SCH<sub>2</sub>,  $J$  = 7.4 Hz), 2.52 d.d (1H, CH<sub>2</sub>,  $J$  = 14.0, 7.2 Hz), 3.60–4.00 m (1H, OH), 3.80–4.20 m (4H, CH<sub>2</sub>O), 4.54 d.d (1H, 8-H,  $J$  = 7.2, 3.8 Hz). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 14.08 (CH<sub>3</sub>), 26.64 (SCH<sub>2</sub>), 44.88 (C<sup>9</sup>), 65.54 (CH<sub>2</sub>O), 65.66 (CH<sub>2</sub>O), 73.51 (C<sup>8</sup>), 115.46 (C<sup>5</sup>), 135.47 (C<sup>6</sup>), 143.58 (C<sup>7</sup>).

## REFERENCES

1. Akbutina, F.A., Torosyan, S.A., and Miftakhov, M.S., *Russ. J. Org. Chem.*, 2001, vol. 37, no. 3, pp. 356–358.
2. Donaldson, R.E. and Fuchs, P.L., *J. Am. Chem. Soc.*, 1981, vol. 103, no. 8, pp. 2108–2110.
3. Donaldson, R.E., Saddler, J.C., Byrn, S., McKenzie, A.T., and Fuchs, P.L., *J. Org. Chem.*, 1983, vol. 48, no. 13, pp. 2167–2188.
4. Johnson, C.R. and Pennig, T.D., *J. Am. Chem. Soc.*, 1988, vol. 110, no. 14, pp. 4726–4735.
5. Akhmetvaleev, R.R., Imaeva, L.R., Belogaeva, T.A., and Miftakhov, M.S., *Izv. Ross. Akad. Nauk, Ser. Khim.*, 1997, no. 9, pp. 1699–1701.
6. Tolstikov, G.A., Ismagilov, S.A., Vel'der, Ya.L., and Miftakhov, M.S., *Zh. Org. Khim.*, 1991, vol. 27, no. 1, pp. 90–95.
7. Miftakhov, M.S., Khalikov, R.M., Akhmetvaleev, R.R., and Tolstikov, G.A., *Russ. J. Org. Chem.*, 1995, vol. 31, no. 2, pp. 182–187.